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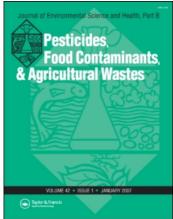
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Journal of Environmental Science and Health, Part B

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597269

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Online Publication Date: 01 August 2009

To cite this Article Beier, Ross C., Anderson, Robin C., Krueger, Nathan A., Edrington, Thomas S., Callaway, Todd R. and Nisbet, David J.(2009)'Effect of nitroethane and nitroethanol on the production of indole and 3-methylindole (skatole) from bacteria in swine feces by gas chromatography', Journal of Environmental Science and Health, Part B,44:6,613 — 620

To link to this Article: DOI: 10.1080/03601230903000701 URL: http://dx.doi.org/10.1080/03601230903000701

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DOI: 10.1080/03601230903000701



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Indole and 3-methylindole (skatole) are odor pollutants in livestock waste, and skatole is a major component of boar taint. Skatole causes pulmonary edema and emphysema in ruminants and causes damage to lung Clara cells in animals and humans. A gas chromatographic method that originally used a nitrogen–phosphorus detector to increase sensitivity was modified resulting in an improved flame ionization detection response for indole and skatole of 236% and 207%, respectively. The improved method eliminates the large amount of indole decomposition in the injector. A 10 μ g mL⁻¹ spike of indole and skatole in water and swine fecal slurries resulted in recovery of 78.5% and 96% in water and 76.1% and 85.8% in fecal slurries, respectively. The effect of the addition of nitroethane and nitroethanol at 21.8 mM in swine fecal slurries was studied on the microbial production of indole and skatole. Nitroethane and nitroethanol decreased the production of skatole in swine fecal slurries at 24 h. The nitroethane effect on L-tryptophan-supplemented fecal slurries after 6 and 24 h incubation resulted in a decrease of 69.0% (P = 0.02) and 23.5% skatole production, respectively, and a decrease of 14.9% indole at 6 h, but an increase in indole production of 81.1% at 24 h.

Keywords: Boar taint; fecal bacteria; indole; metabolism; nitroethane; nitroethanol; pulmonary edema; skatole; swine feces; tryptophan.

Introduction

Sweeping changes have been made in livestock production over recent years that have resulted in intensive production techniques generating large amounts of liquid manures.^[1] The storage and land application of livestock manure can cause an odor nuisance, which may result in the reduction in value of neighboring properties.^[2] Various odorants have been identified as being important contributors to livestock odor nuisance.^[3] Commonly, ammonia and hydrogen sulfide emissions are evaluated (for example^[4]). However, indole and 3-methylindole (skatole) (Fig. 1, A

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and B) have also been identified in the odorous emissions from pig slurries.^[2,3,5] Skatole was one of the major volatile components in pig slurry odor and was second in quantity to 4-methylphenol in a pig slurry diethyl ether extract.^[3] High levels of skatole in the fat of male pigs along with high levels of the steroid pheromone, androstenone, is the major cause of boar taint, the unpleasant flavor associated with meat from male pigs.^[6–11] Indole and skatole are the main products of bacterial L-tryptophan metabolism,^[12] and maybe as important, indole and skatole are produced from L-tryptophan by bacteria in bovine rumen.^[13,14]

Intraruminal doses of L-tryptophan caused acute pulmonary edema and emphysema in cattle.^[15–17] The L-tryptophan-induced pulmonary edema and emphysema in cattle was associated with ruminal and plasma concentrations of skatole, and severity of the pulmonary lesions were related to the concentration of skatole found in the plasma.^[18] Skatole has been associated with a number of disease conditions in various animal species. Pulmonary edema and emphysema was demonstrated in cattle following intraruminal and intravenous administration of skatole.^[20] Oral or intravenous administration of skatole causes destruction

Fig. 1. Chemical structures of (A) indole, (B) 3-methylindole (skatole), and (C) 3-methyleneindolenine (the reactive intermediate causing pneumotoxicity).

of lung Clara cells in goats^[21,22] and sheep^[23] as well as selective damage to lung Clara cells in the horse,^[24] rat and mouse.^[25] Skatole also causes olfactory mucosal injury in the mouse,^[26] disruption of axonal transport between the olfactory epithelium and olfactory bulb in the rat,^[27,28] and olfactory bulb deafferentation in the rat.^[29]

The toxicity is caused by a transient free radical^[30] produced by cytochrome P-450 bioactivation of skatole.[31] The toxic intermediate from bioactivated skatole has been shown to be the methylene imine, 3-methyleneindolenine (Fig. 1, C), [32] which covalently binds proteins. [32,33] Bioactivation of skatole resulting in toxicity is similar to the cytochrome P-450 bioactivation of two other toxic natural products. The skatole-induced pulmonary lesions in mice^[26] resemble the sweet potato lung toxin, 4-ipomeanol, induced lesions in rodents.^[34] Cytochrome P-450 bioactivates 4-ipomeanol in lung Clara cells, the reactiveintermediate readily covalently binds proteins, and is known to cause lung edema in cattle.[35] Goats are the most susceptible animal to skatole toxicity, [20,32] and goats are also very susceptible to white snakeroot toxicity.[36] White snakeroot contains tremetone that can be bioactivated by cytochrome P-450 resulting in cell killing in cell culture. [37,38] But white snakeroot results in general tissue toxicity, and not only primarily in lung tissue. [36]

Humans may be susceptible to skatole-mediated toxicities. Human bronchial epithelial cells bioactivate skatole to produce 3-methyleneindolenine, which induces programmed cell death at low skatole concentrations. Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additional routes of exposure to skatole; cheeses, Humans may have additiona

workers experience significant insult to the respiratory tract resulting in bronchitis and reactive airways. Neighbors of a large-scale swine operation reported experiencing higher rates of symptoms associated with respiratory inflammation than did demographically similar subjects living near minimal livestock production. Environmental odors emanating from large-scale hog operations may also have a pronounced effect on mood and stress levels. Previous observations suggested that nitroethane and nitroethanol may reduce the odor of swine feces (R.C.A. personal communication). To investigate ways of altering microbial degradation of L-tryptophan in fecal slurries, it was necessary to develop a simple, accurate method to quantify indole and skatole.

Various methods have been used to quantify indole and skatole. Skatole has been quantified from pig backfat by gas chromatography (GC),^[53,54] high-performance liquid chromatography (HPLC),^[55] and spectrophotometry methods. [8,56] HPLC was used to quantify indole and skatole in swine gut contents,^[57] and to quantify skatole in various animal feces.^[58] A GC-mass selective detector (MSD) was used to determine the emission rates of indole, skatole, and other odorous compounds from pig fecal slurries.^[5] GC-mass spectrometry (MS) was used to quantify indole and skatole in the volatile compounds from poultry manure, [59] in pig and chicken waste odors, [3] and in emissions of swine waste lagoons. [60] GC has been used to quantify indole and skatole from bovine plasma^[61] and ruminal fluid.[17] A GC method using a nitrogen-phosphorus detector to improved sensitivity was used to quantify indole and skatole from culture media, intestinal contents, feces, [62] and in pig fecal slurries. [63]

The initial goal of the current study was to develop a simple GC method for the quantification of indole and skatole in pig fecal slurries. The conditions used by Jensen and Jensen^[62,63] were studied to determine the source of poor sensitivity that was obtained for indole and skatole during extraction and GC analysis that required the use of a nitrogen–phosphorus-sensitive detector. In this study, a previous method^[62] was modified resulting in an improved flame ionization detector (FID) response. The second goal was to use this new GC-FID method to evaluate the effect of nitroethane and nitroethanol on the production of indole and skatole in swine fecal slurries.

Materials and methods

Chemicals and materials

Indole, 5-methylindole, NaCl, nitroethane, nitroethanol, skatole, and 200 proof anhydrous EtOH were obtained from Sigma–Aldrich (Milwaukee, WI). L-Tryptophan was obtained from Sigma–Aldrich (St. Louis, MO). Hexanes (boiling range 68–70°C) was obtained from EM Science (Gibbstown, NJ). Five-mL sterile microtubes

(No. 80078-650) were obtained from VWR International (West Chester, PA). Whirl-packs were obtained from BVA Scientific (San Antonio, TX). H₂O was produced on site by a reverse osmosis system obtained from Millipore Corp. (Bedford, MA).

Sample collection

Swine feces were collected from market aged feeder pigs immediately following voiding. The collected feces was placed in whirl-packs and delivered to the laboratory (adjacent to the farm site). A swine fecal slurry was then generated by making a 1:5 mixture of swine feces and an anaerobic dilution solution, respectively, according to Bryant and Burkey. The feeder pigs were raised according to standard husbandry practices as defined by our Institutional Animal Care and Use Guidelines, No. 2007001.

Instrumentation

The Jensen and Jensen^[62] gas chromatographic method was modified by using a 25 m \times 0.32 mm ID, BP20 (0.5 μ m film thickness) capillary column (SGE Inc., Austin, TX) on a HP 6890 GC System (Agilent Technologies, Santa Clara, CA) with FID peak detection. Hydrogen was used as the carrier gas at 2.0 mL min⁻¹. The mode was splitless and the injector temperature was set at 250°C with an initial column temperature of 80°C. A double ramp temperature program was used at 16°C min⁻¹ to 185°C, hold for 6.25 min, followed by 20°C min⁻¹ to 240°C, with a final hold for 16.3 min. The Hermle Z230A centrifuge was obtained from National Labnet Co. (Woodbridge, NJ).

Analytical procedures

Preparation of standards

Indole and skatole standards (200 μg mL⁻¹) used for producing the standard curve were made in hexanes (boiling range 68–70°C). The 5-methylindole internal standard (ISD) (200 μg mL⁻¹) was made in 200 proof anhydrous EtOH. Standard curves were prepared for indole and skatole by adding quantities of indole and skatole standards in hexanes with 30 μ L of the ISD solution plus hexanes solvent to produce a concentration range of 0.5–90 μg mL⁻¹ indole and skatole with a concentration of 6 μg mL⁻¹ ISD. Indole and skatole 200 μg mL⁻¹ standards used for spiking water samples and fecal slurries were made in 200 proof anhydrous EtOH.

Injector temperature experiment

Indole and skatole samples in hexanes solvent ($10 \,\mu g \, mL^{-1}$) were injected into a 250°C injector, and the FID response in pA was used for comparison to indole and skatole samples in chloroform solvent ($10 \,\mu g \, mL^{-1}$) injected into a 150, 200, 250, 275, and 300°C injector.

Extraction method

Samples were quantified for indole and skatole using the following extraction method: To a 1 mL water spiked or fecal slurry sample, a 1 mL NaCl solution (34 g NaCl per 100 mL H₂O) was added in a 5-mL sterile microtube. To this solution 50 μ L of 4N NaOH and 30 μ L of ISD (6 μ g mL $^{-1}$ final concentration) were added. Finally, 1 mL of hexanes was added and the sample was vortexed until an emulsion formed, and then centrifuged for 10 min at 2,739 \times g in a Hermle Z230A centrifuge. The hexanes layer was then transferred to a sample vial and a 2 μ L sample was injected into the GC.

Water and fecal slurry spiking experiment

Water was spiked with indole and skatole EtOH standards resulting in a final concentration of $10~\mu g~mL^{-1}$. These spiked water samples were then extracted and analyzed by GC according to the above methods. The swine fecal slurry was spiked with indole and skatole standards in EtOH resulting in a final concentration of $10~\mu g~mL^{-1}$ of either compound. These spiked fecal slurry samples were also extracted and analyzed by GC according to the above methods.

Nitroethane and nitroethanol effect on indole and skatole production

The swine fecal slurry was anaerobically (under 100% CO₂) distributed in 12 mL volumes. Nitroethane or nitroethanol (0.1 mL, 2664 mM in H₂O) was anaerobically added to the appropriate sample tubes (final concentration = 21.8 mM) with and without L-tryptophan (0.1 mL, 24 mg mL⁻¹ in H₂O). H₂O (0.1 or 0.2 mL) was added anaerobically to the appropriate tubes to make the mixtures have equal volume. Supplemented fecal slurries and other controls were kept anaerobically until samples (1 mL) were processed according to the extraction method above at 0.5, 6, and 24 h, and then indole and skatole were quantified according to the above GC method.

Data analysis

Concentrations of indole and skatole during in vitro incubations were analyzed for treatment differences by a repeated measures analysis of variance using Statistix8 Analytical Software (Tallahassee, FL). Means were separated using a Tukey's procedure.

Results and discussion

The analysis of indole and skatole (Fig. 1, A and B) required a simple, effective analysis method. The method of Jensen and Jensen^[62] initially appeared to be appropriate; however, it was quickly realized that the method resulted in extremely poor sensitivity for indole and skatole using FID. Thus the method^[62] was investigated to determine if it

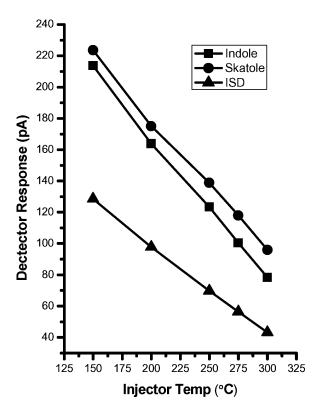


Fig. 2. Detector response for indole, skatole, and the internal standard (ISD) with CHCl₃ as the injection solvent at injector temperatures of 150–300 °C.

might be modified to increase the sensitivity using FID. The original method used chloroform for indole and skatole extraction and as injection solvent in the GC at an injector temperature of 300°C. Figure 2 shows the results of injecting a 10 μ g mL⁻¹ chloroform solution of indole and skatole containing the ISD 5-methylindole in the GC with injector temperatures ranging from 150–300°C. Detector responses for all three chemicals were drastically reduced as the injector temperature increased to 300°C. This loss in sensitivity using chloroform as solvent led to Jensen and Jensen^[62] requiring the use of a nitrogen–phosphorus-sensitive detector, but the large decomposition of these indoles in the injector could alter the value of the results.

To improve the sensitivity of the Jensen and Jensen method, the method was modified in three ways: a) hexanes (boiling range $68-70^{\circ}$ C) was used instead of chloroform for extraction and injection solvent, b) the injector temperature was lowered to 250° C, and c) a salt solution was added during extraction instead of only H₂O. These changes resulted in an FID response for indole, skatole, and ISD of 185.1 ± 2.8 , 198.4 ± 2.9 , and 117.5 ± 1.8 pA, respectively. In comparison with the method of Jensen and Jensen^[62] (Fig. 2, 300° C injector temperature with chloroform solvent), the detector response for indole, skatole, and the ISD using the modified method increased the FID response by 236, 207, and 272%, respectively, and provided

sufficient sensitivity for quantification of indole and skatole. Standard curves for indole and skatole with the new modified method had a correlation of r=0.99984 and 0.99988, respectively. A 10 μg mL⁻¹ spike of indole and skatole in water and a fecal slurry is shown in comparison

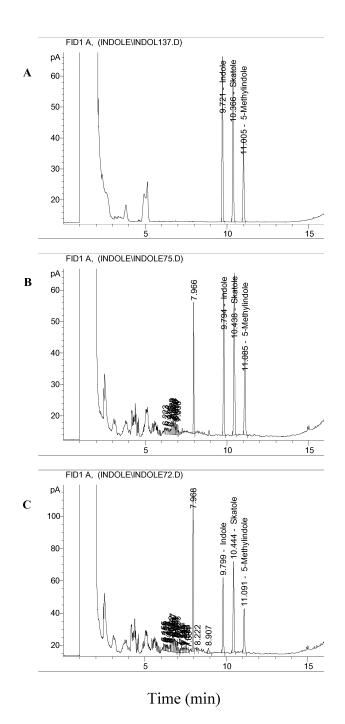


Fig. 3. Gas Chromatography (GC) tracings of (A) $10~\mu g~mL^{-1}$ indole and skatole with the internal standard, 5-methylindole, in hexanes; (B) a $10~\mu g~mL^{-1}$ spike of indole and skatole in water, and (C) a spike of $10~\mu g~mL^{-1}$ indole and skatole into a pig fecal slurry followed by the described extraction and chromatography methods.

Table 1. Recovery of $10 \mu g \text{ mL}^{-1}$ indole and skatole spikes in water and control swine fecal slurries.*

	$10~\mu g~mL^{-1}$ spike in water $(\mu g~mL^{-1})^{\dagger}$	% recovery	10 $\mu g \ mL^{-1}$ spike in fecal slurries ($\mu g \ mL^{-1}$) †	% recovery
Indole Skatole	$7.85 \pm 0.22 \\ 9.60 \pm 0.31$	78.5 96.0	$7.61 \pm 0.15 \\ 8.58 \pm 0.18^{\ddagger}$	76.1 85.8

^{*}The initial concentration of indole and skatole in the control swine fecal slurry was <0.5 and $1.66 \pm 0.05 \,\mu \mathrm{g \ mL^{-1}}$, respectively.

to a $10 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$ hexanes solution of indole, skatole, and the ISD (6 $\mu\mathrm{g}\,\mathrm{mL}^{-1}$) (Fig. 3). Recovery of $10 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$ spikes of indole and skatole from water was 78.5% and 96.0%, respectively, and recovery from fecal slurries was 76.1% and 85.8%, respectively (Table 1).

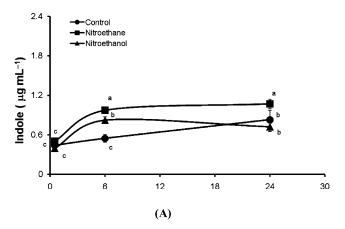
The effect of nitroethane and nitroethanol on the production of indole and skatole in swine fecal slurries over a 24 h period is shown in Fig. 4. At 24 h, the level of skatole (Fig. 4B) was significantly lower than the control. Also, the nitroethane and nitroethanol treatment produced less skatole at 6 h than the control. In these slurries both the nitroethane and nitroethanol additions increased the levels of indole at 6 h (Fig. 4A), and the indole level in the nitroethane treatment continued to be elevated at 24 h.

The addition of L-tryptophan to fecal slurries resulted in a large increase of indole and skatole production, 2578% and 104%, respectively, at 6 h, and 1314% and 204%, respectively, at 24 h. However, addition of nitroethane to the L-tryptophan containing mixtures resulted in a decrease of 14.9% indole and 69.0% skatole at 6 h, and a decrease of 23.5% skatole at 24 h (Fig. 5). But, indole increased by 81.1% at 24 h (Fig. 5A). The effect of nitroethanol on L-tryptophan metabolism after 6 and 24 h incubation resulted in a decrease of 8.2% indole and 41.2% skatole at 6 h and a decrease of 8.8% skatole at 24 h (Fig. 5). Again, indole levels increased by 72.2% at 24 h (Fig. 5A). It appears that the addition of both nitroethane and nitroethanol to L-tryptophan-supplemented fecal slurries decreased the

production of indole and skatole at 6 h, and decreased skatole production at 24 h (Fig. 5). Even though the addition of nitroethane and nitroethanol did not eliminate the production of skatole in L-tryptophan-supplemented fecal slurries, nitroethane decreased the production of skatole (P = 0.02) after 6 h incubation. At 24 h the production of skatole (Fig. 5B) was still somewhat inhibited by both nitroethane and nitroethanol, while the production of indole significantly exceeded the L-tryptophan control values for indole (Fig. 5A).

The type and amount of carbohydrate source has a substantial effect on nitrogen metabolism and may influence the synthesis of indoles. It was previously shown that addition of fructooligosaccharide to L-tryptophan-supplemented swine fecal slurries caused a significant decrease in skatole production and a marginal increase in the rate of indole production. The researchers hypothesized this resulted from a shift in microbial metabolism due to an increased need for amino acids. In our experiments, nitroethane and nitroethanol also appeared to cause a shift in microbial metabolism.

The metabolism of L-tryptophan in bacteria occurs through several pathways.^[67] L-Tryptophan can be metabolized to indole-3-acetamide or indole-3-acetaldehyde, both of which can be further metabolized to indoleacetic acid (IAA), and skatole can be produced from IAA.^[17] But L-tryptophan can be converted to other products, too (e.g., *N*-formyl-kynurenine, indole-3-pyruvate, and tryptamine).



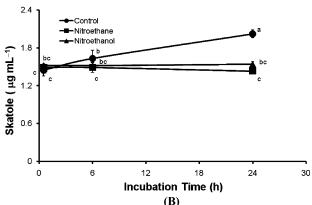


Fig. 4. Concentration of indole (A) and skatole (B) in swine fecal slurries with the addition of nitroethane and nitroethanol compared to the control over a 24 h period.

[†]The result is the mean of 4 spiked samples.

[‡]The value for the initial amount of skatole in the control swine fecal slurry was subtracted from the observed skatole values.

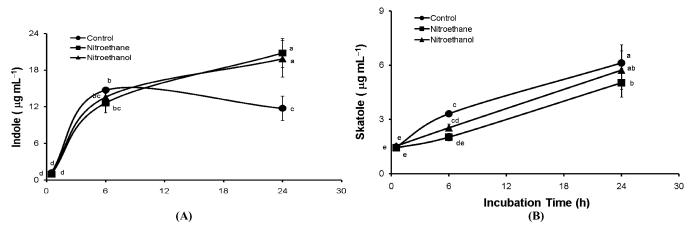


Fig. 5. Concentration of indole (A) and skatole (B) in swine fecal slurries supplemented with L-tryptophan with addition of nitroethane and nitroethanol compared to the control over a 24 h period.

Indole-3-pyruvate can be further metabolized to form IAA.[67] However, only a single metabolism step is required for the production of indole from L-tryptophan in bacteria. [67] Little is known concerning the metabolism of L-tryptophan to skatole in feces, but in the rumen Ltryptophan is metabolized by at least two different pathways to produce either indole, by the single step conversion mentioned earlier, or skatole. [68] *Lactobacillus* sp. were isolated from rumen that could produce skatole from IAA.[13] These Lactobacillus sp. could not produce indole, IAA, or skatole directly from L-tryptophan. The ability of ruminal Lactobacillus sp. to produce skatole from IAA appeared to be metabolically unique.^[69] It was communicated that IAA could be converted to skatole in a variety of animal fecal microflora, including pigs and humans. [69,14] Skatole inhibited the growth of and was toxic to the Lactobacillus sp. that produced it and was a non-competitive inhibitor of the skatole-forming enzyme.^[70] Also, some rumen bacteria can produce L-tryptophan from IAA.^[71] Therefore, since Ltryptophan can be formed from intermeadiates like IAA,[71] the inhibition of skatole production may result in more Ltryptophan being available for indole production. We hypothesize that nitroethane and nitroethanol are inhibiting the pathway from L-typtophan to skatole; therefore, allowing the indole pathway to proceed. The question remains, what reason or benefit does the production of skatole have for the bacteria?^[72]

Conclusions

The analytical method presented here for the quantification of indole and skatole in swine fecal slurries is a simple and quick method for determining the levels of these chemicals. Modifications to the method of Jensen and Jensen^[62] by the addition of saline instead of water during the extraction process, use of hexanes (68–70°C fraction) for extraction and injection solvent, and decreasing the injector temperature to 250°C improved the FID response, which

allowed the modified method to be successful. The improved method eliminates the large amount of indole and skatole decomposition in the injector. The new improved method resulted in a recovery of 78.5% indole and 96.0% skatole from water, and a recovery of 76.1% indole and 85.8% skatole from fecal slurries. These results suggest that perhaps indole and skatole are binding or reacting with a fecal slurry component(s).

This work demonstrates that a gas chromatographic FID detector can be used to quantify indole and skatole, and that certain additives like nitroethane and nitroethanol can alter the production of indole and skatole from bacteria in swine fecal slurries and from L-tryptophan-supplemented swine fecal slurries. Nitroethane and nitroethanol reduced the level of skatole in swine fecal slurries and tended to decrease the level of skatole in L-tryptophan-supplemented swine fecal slurries. The levels of indole significantly increased in L-tryptophan-supplemented swine fecal slurries upon addition of either nitroethane or nitroethanol. The level of indole also increased at 6 and 24 h when nitroethane was added to swine fecal slurries. Nitroethanol also increased the level of indole over the control at 6 h. Suggesting that addition of nitroethane or nitroethanol may be inhibiting the pathway from L-tryptophan to skatole, allowing the indole pathway to proceed. Based on this work, further studies can be made to evaluate inhibitors of the production of indole and skatole in fecal slurries. Due to the lung Clara cell toxic effects caused by skatole, the reduction or elimination of skatole production and accumulation in animal waste storage pits, waste holding reservoirs or lagoons may be beneficial to farm workers and especially to swine confinement workers.

Acknowledgments

We thank Jackie Kotzur for technical assistance. This work was funded by the United States Department of Agriculture, Agricultural Research Service.

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